This article was downloaded by: [University of Haifa Library]

On: 17 August 2012, At: 19:37 Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH,

UK



# Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information: <a href="http://www.tandfonline.com/loi/gmcl19">http://www.tandfonline.com/loi/gmcl19</a>

Anomalies within Alkyl and Alkoxy Substituted Bis (1,3-diphenylpropane - 1,3 dionato) Copper (II) metallomesogens from X-ray and EPR Studies

Monisha Bose <sup>a</sup> , Kazuchika Ohta <sup>b</sup> , Tosio Sakurai <sup>c</sup> & Chanchal K. Majumdar <sup>a</sup>

<sup>a</sup> S. N. Bose National Centre for Basic Sciences, Block JD, Sector III, Salt Lake, Calcutta, 700091, India

<sup>b</sup> Dept. of Functional Polymer Science, Faculty of Textile Science and Technology, Shinshu University, UEDA 386, Japan

<sup>c</sup> Department of Science, Faculty of Education, Shinsu University, Nagano, 381, Japan

Version of record first published: 24 Sep 2006

To cite this article: Monisha Bose, Kazuchika Ohta, Tosio Sakurai & Chanchal K. Majumdar (1999): Anomalies within Alkyl and Alkoxy Substituted Bis (1,3-diphenylpropane - 1,3 dionato) Copper (II) metallomesogens from X-ray and EPR Studies, Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals, 326:1, 229-247

To link to this article: <a href="http://dx.doi.org/10.1080/10587259908025417">http://dx.doi.org/10.1080/10587259908025417</a>

### PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <a href="http://www.tandfonline.com/page/terms-and-conditions">http://www.tandfonline.com/page/terms-and-conditions</a>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

# Anomalies within Alkyl and Alkoxy Substituted Bis (1,3-diphenylpropane - 1,3 dionato) Copper (II) metallomesogens from X-ray and EPR Studies

MONISHA BOSE a, \*, KAZUCHIKA OHTAb, TOSIO SAKURAI<sup>c</sup> and CHANCHAL K. MAJUMDAR a

(Received 9 September 1997; In final form 15 June 1998)

Our previous EPR studies dealt with the contrasting behaviour of the lamellar C<sub>8</sub>OCu, as against the columnar  $C_8Cu/C_{10}Cu$ . X-ray and EPR studies of  $C_nCu(n = 6 - 12)$  and  $C_nOCu(n = 6 - 12)$  presented here, however, indicate that some members exhibit anomalous behaviour within each series. Thus C<sub>6</sub>Cu, C<sub>8</sub>Cu, C<sub>10</sub>Cu and C<sub>12</sub>Cu (even members) show a single exchange-narrowed strong asymmetric  $g_{\perp}$  along with a weak  $g_{\parallel}$  in pure solids. Interestingly,  $C_7Cu$  and  $C_9Cu$  (odd members) show hyperfine quartets in  $g_{\parallel}$ , indicating weak or no exchange.  $C_7Cu$  shows a single  $g_{\perp}$ , conforming to the axial symmetry in  $C_nCu$ . However, C<sub>9</sub>Cu is rather unique: gives three different mesophases and exhibits non-axial behaviour. Crystal structure for C<sub>6</sub>Cu and C<sub>7</sub>Cu reported here for the first time, indicate that the former has a structure similar to that of  $C_8Cu$ . However,  $C_7Cu$ , which also has z = 1 and belongs to the same triclinic space group, has a close interaction between two molecules at the opposite corners of the ab plane in the unit cell, through chain - chain intermolecular interaction and possibility  $\pi - \pi$  interaction among phenyl-rings in adjacent molecules. This is reflected in the 'c' parameter, which is 10.066 Å, as compared to 5.863 Å in C<sub>6</sub>Cu. However, this interaction in C<sub>7</sub>Cu vanishes in the mesophase, as indicated by the collapse of the quartet. Weak exchange in C<sub>7</sub>Cu is also reflected in the proton relaxation time  $T_1$  (68.9 msecs), as compared 1.9 msecs for  $C_6Cu$ . In  $C_nOCu$ ,  $C_7OCu$  behaves like the dimer  $C_8OCu$ , with a complex hyperfine structure including quadrupole forbidden transitions after  $g_1$ , along with a broad  $g_2$  and a weak  $g_3$ . The dimer in C<sub>8</sub>OCu is strong and the quartet persists in the mesophase. Other members, possibly monomers viz., C<sub>6</sub>OCu, and C<sub>10</sub>OCu, give a hyperfine quartet in  $g_1$  (while in C<sub>12</sub>OCu it is after  $g_1$ ),

<sup>&</sup>lt;sup>a</sup> S. N. Bose National Centre for Basic Sciences, Block JD, Sector III, Salt Lake, Calcutta-700091, India;

<sup>&</sup>lt;sup>b</sup> Dept. of Functional Polymer Science, Faculty of Textile Science and Technology, Shinshu University, UEDA 386, Japan;

<sup>&</sup>lt;sup>c</sup> Department of Science, Faculty of Education, Shinsu University, Nagano 381, Japan

<sup>\*</sup>Corresponding author. Tel.: 321-5705-10, Fax: 91-33-334-3477, e-mail: root@bose.ernet.in

followed by a strong single  $g_2$  line and a weak  $g_3$ , conforming to the non-axial symmetry of the lamellar complex. Thus, chain length does play a role but not in a linear fashion, because of the anomalies presented above.

Keywords: Copper metallomesogen; X-ray; EPR

### 1. INTRODUCTION

The first paramagnetic disc-like metallomesogen bis[1,3-di(p-n-decylphenyl) propane-1,3 dionato] copper(II) was reported by Giroud-Godquin and Billard [1]. Levelut [2] from X-ray studies of this compound (abbreviated as C<sub>10</sub>Cu), considered it to have a lamellar phase with imperfect crystalline organisation. Billard [3] concluded from polarizing microscopic observations that the mesophase of  $C_{10}$ Cu has both a columnar and layered (lamellar) structure. From binary phase diagram studies of the homologues viz., C<sub>8</sub>Cu and C<sub>10</sub>Cu and the corresponding alkoxy complex C<sub>8</sub>OCu, Ohta et al. [4, 5] showed that each of these exhibit mesomorphic phases. C<sub>8</sub>Cu had two different mesophases  $D_1$  and  $D_2$ . The latter phase in  $C_8$ Cu is the same mesophase of Giroud-Godquin and Billard, while the D phase in C<sub>8</sub>OCu was completely miscible with the D<sub>1</sub> Phase of C<sub>8</sub>Cu. Further, Ohta et al. [6] and Sakashita et al. [7] revealed from X-ray studies, that the D mesophase in C<sub>n</sub>OCu has a lamellar structure without columnar stacking and the molecules tilt 5° to the layer. Ohta et al., denoted these  $D_1$  and  $D_2$  phases as  $D_{L1}$  (discotic lamellar mesophases without columnar structure) and  $D_{L2}$  (discotic lamellar mesophases with columnar structure). Thus, these  $D_{L1}$  and  $D_{L2}$  mesophases are very unique and different from conventional discotic columnar and smectic mesophases.

To study the effect of chain length on mesomorphic properties, Ohta et al. [6, 8] had synthesised the whole series of  $C_nCu$  and  $C_nOCu$  complexes (Fig. 1) from n = 0-12, most of which showed two solid polymorphs. Recently, Usha et al. [9] have shown that  $C_8Cu$  also exhibits two solid polymorphs (needle-like and prismatic) with very different crystal structures. Only  $C_{12}Cu$ , first reported by Giroud-Godquin et al. [10] does not exhibit any polymorphism. In the  $C_nCu$  series, complexes from n = 4-12 were found to be mesogenic. Some members show plural mesophases viz.,  $C_7Cu$  and  $C_8Cu$  show two different mesophases  $D_1$  and  $D_2$ , whereas  $C_9Cu$  shows three mesophases  $D_1$ ,  $D_2$  and  $D_3$ . Ohta et al. [8] have shown that the textures of  $D_1$  and  $D_2$  phases are the same in  $C_7Cu$  and  $C_8Cu$ .  $D_1$  in  $C_8Cu$  gave a 'plane' texture surrounded by a lustrous ring. In  $D_2$ , this structure got wrinkled.  $C_9Cu$  was exceptional with three mesophases:  $D_1$  showed a spiral

R
$$R = C_nH_{2n+1}$$
 or  $C_nH_{2n+1}$ 
where  $n = 6 - 12$ 

FIGURE 1 Structure of alkyl/alkoxy substituted aryl  $\beta$  diketonates of Cu(II).

texture, which was destroyed in  $D_2$  to give a mosaic texture, surrounded by a lustrous ring. In  $D_3$ , this ring began to be distorted [8] and the mosaic texture began to wrinkle. Thus within this series, the properties differ greatly and abruptly.

In Comparison,  $C_nOCu$  appears to be less complex. Thus, all the mesogens from n = 6-12 give a single mesophase [6], with big broken fan-like texture. Unlike the  $C_nCu$  series, the crystal shape is rod-like for  $C_6OCu(K_2)$ ,  $C_7OCu(K_2)$  and  $C_8OCu(K_3)$  but plate-like for  $C_9OCu(K_3)$ ,  $C_{10}OCu$  and  $C_{11}OCu$ . However,  $C_{12}OCu$  is needle-like as in  $C_nCu$ . All of them give a lamellar mesophase  $(D_{L1})$ , inspite of differences in their molecular shape and may be in crystal structure. Except for  $C_8OCu$ , crystal structures of  $C_nOCu$  complexes could not be determined, as single crystals could not be grown. Each of the complexes from  $C_6OCu$  to  $C_{12}OCu$  could easily afford a supercooled discophase at room temperature, which indicates, that this phase  $(D_{L1})$  in  $C_nOCu$  has a rather stable structure, and possibly is the reason, why  $C_nOCu$  does not have any plural mesophase.

The point to note is that in both the series, though the different polymorphs of a complex give different crystal structures having different melting points (MP), they give the same mesophase. The transition from K to the discotic mesophase involves a significant enthalpy change ( $\Delta H$ ). In the  $C_nCu$  series, enthalpy for melting is larger than the enthalpy for clarification, but in  $C_8OCu$ , the enthalpy of clarification is higher than that

TABLE I Phase Transition Temperatures  $(T_i)$  and enthalpy changes  $(\Delta H_i)$  n versus  $T_i(^{\circ}C)/\Delta H_i(\text{kcal/mol})$ 

| Columnar $C_nCu$ $(n=6-12)$   | Mesophase | :      | Lamellar $C_nOCu$ $(n = 6-12)$  |
|---|-----------|--------|---|
| $C_6Cu(K_1)\frac{79.3}{13.4}D\frac{154.0}{6.33}H$   | One       |        | $C_6OCu(K_2) \xrightarrow{152.9} D \xrightarrow{182.4} IL$                          |
| $(\mathbf{K}_2)\frac{91.8}{5.53}$   |           |        | $(\mathbf{K}_1) \frac{125.7}{1.0}$  |
| $C_7 \text{Cu}(\mathbb{K}_2) \frac{76.2}{13.71} D_1 \frac{94.6}{0.84} D_2 \frac{150.0}{6.92} IL$    | Two       |        | $C_7OCu(K_2) \xrightarrow{111.2}_{12.8} D \xrightarrow{180.8}_{12.1} IL$            |
| $(\mathbf{K}_1)_{\underline{53.9}}$   |           |        | $(\mathbf{K}_1)\frac{93.9}{3.9}$  |
| $C_8Cu(K) \frac{76.1}{26.80} D_1 \frac{117.2}{0.76} D_2 \frac{141.6}{8.60} IL$                      | Two       | Single | $C_8OCu(K_3) \xrightarrow{93.1} D \xrightarrow{173.3} L$                            |
|   |           |        | $(\mathbf{K}_1) \xrightarrow{21.8}^{21.8} (\mathbf{K}_2) \xrightarrow{82.5}^{82.5}$ |
| $C_9Cu(K_1)\frac{80.0}{21.66}D_1\frac{97.8}{1.20}D_2\frac{111.1}{0.49}D_3\frac{137.0}{7.47}H$       | Three     | Meso-  | $C_9OCu(K_3) \xrightarrow{94.5} D \xrightarrow{170.7} IL$                           |
| $(\mathbf{K}_2)\frac{82.6}{2.39}$   |           |        | $(\mathbf{K}_1)\frac{55.6}{3.1}(\mathbf{K}_2)\frac{84.2}{0.5}$                      |
| $C_{10}Cu(K_2) \stackrel{91.3}{\underset{29.50}{\sim}} D \stackrel{130.1}{\underset{7.71}{\sim}} L$ | One       | phase  | $C_{10}OCu(K) \xrightarrow{81.9} D \xrightarrow{166.6} IL$                          |
| $(\mathbf{K}_1) \frac{860}{21.67}$  |           |        |   |
| $C_{11}Cu(K_1) \stackrel{842}{=} D \stackrel{1253}{=} L$  | One       |        | $C_{11}OCu(\mathbf{K}) \xrightarrow{87.6} D \xrightarrow{161.2} IL$                 |
| $(\mathbf{K}_2)^{\underline{56.6}}$   |           |        |   |
| $C_{12}Cu(K) \xrightarrow{98.9} D \xrightarrow{119.8} IL$   | One       |        | $C_{12}OCu(K) \xrightarrow{74.3} D \xrightarrow{154.1} L$                           |

of fusion, which is a rare situation for organic discogens, first reported by Fugnitto *et al.* [11]. However in  $C_8OCu$ , this situation is now understandable from Usha *et al.*'s [12] crystallographic studies. In this compound, the repeating unit is a dimer with z=2, as against the monomer in  $C_8Cu$  with z=1. The strong bond in the dimer persists in the mesophase (high enthalpy not required) but is broken only in the isotropic phase (higher enthalpy required), as has been shown from EPR [15].

Eastman et al. [13] first reported that single crystals of C<sub>8</sub>Cu show a single exchange-narrowed EPR line in X-band at room temperature. He considered C<sub>8</sub>Cu to be a one-dimensional Heisenberg antiferromagnet and could observe the phase transition from K to the D<sub>1</sub> phase. The change in spectrum was observable, but not dramatic, as exchange effects still persisted in the mesophase. In C<sub>8</sub>OCu, they did not see the exchange-narrowed line but observed a complex spectrum with a hyperfine structure. No spectrum of C<sub>8</sub>OCu was presented and no further work was reported. This motivated Bose and Sadashiva [14, 15] to pursue a thorough study of this contrasting behaviour between alkyl and alkoxy substitution in aryl  $\beta$  diketonates of Cu from X band and Q band EPR. Their early <sup>1</sup>H NMR studies [14] in C<sub>10</sub>Cu and C<sub>8</sub>OCu, though confirmed fluidity in these discogens, failed to distinguish between columnar C<sub>10</sub>Cu and lamellar C<sub>8</sub>OCu. However, they have shown from magnetic susceptibility studies [16] that Eastman's contention of C<sub>8</sub>Cu being a spin 1/2 one dimensional Heisenberg antiferromagnet at room temperature is not true, as C<sub>8</sub>Cu fails to order even at liquid heluim temperatures. But the strong antiferromagnetic exchange  $(J_{ex} \gg A)$ in C<sub>8</sub>Cu/C<sub>10</sub>Cu is confirmed from the strong exchange-narrowed asymmetric  $g_{\perp}$  line (14), along with a very weak  $g_{\parallel}$  (axially symmetric). C<sub>8</sub>OCu in the Q band on the other hand, clearly revealed a complex hyperfine pattern [15] after a weak  $g_1$ , along with a wide (140 Oe)  $g_2$  line and a weak  $g_3$ , indicating a non-axial system, with weak or no exchange-narrowing  $(J_{\rm ex} \ll A)$ . Thus C<sub>8</sub>Cu EPR, conformed to the axial columnar system and C<sub>8</sub>OCu to the non-axial staggered lamellar structure, as reported in X-ray studies by Usha et al. [12, 17]. Q band spectra of C<sub>8</sub>OCu, indicated the presence of quadrupole forbidden transitions [18], along with the Cu hyperfine quartet from the distorted dimer of C<sub>8</sub>OCu, the repeating unit in the unit cell. The copper ion was not on an inversion centre, as in C<sub>8</sub>Cu, but in a zig-zag configuration, with Cu-Cu distances alternating between 6.3 and 6.6 Å. The existence of quadrupolar forbidden transition has been confirmed by a computer simulation of the hyperfine spectrum and good agreements between line positions and intensites were obtained from C<sub>8</sub>OCu and the mixed complex  $2C_8Cu-2OC_8$  [16].

However, when  $C_8Cu$  and  $C_{10}Cu$  were doped into the corresponding Pd complex, the weak  $g_{\parallel}$  splitted up to give a hyperfine quartet, in which the outer lines suffered isotopic splitting in the ratio of 1:2, due to presence of  $^{35}Cu$  and  $^{37}Cu$  in that ratio [19]. The corresponding  $g_{\perp}$  line gave a hyperfine structure with Cu quartet and quadrupole forbidden transitions in place of the single exchange-narrowed line. Similar spectra were obtained with  $C_8Cu/C_{10}Cu$  in frozen chloroform solution. Thus, at the molecular level,  $C_8Cu$  and  $C_8OCu$  behave in the same fashion. However, in the solid, packing effects dominate.  $C_8Cu$  with Cu-Cu distances at 5.82 Å gives a one-dimensional close-packed columnar structure, wherein strong exchange-interaction wipe out any fine structure, whereas in  $C_8OCu$ , the bulky alkoxy group lead to a staggered two-dimensional lamellar structure with Cu-Cu distances > 6 Å, leading to absence of long-range exchange-interaction effects.

In the present paper, X-ray and EPR studies have been extended to cover the two entire series,  $C_nCu$  and  $C_nOCu$ , in an effort to unravel any correlation which may exist between polymorphism, plural mesophase formation, texture and crystal structure with changes in chain length across the series.

### 2. EXPERIMENTAL

# 2.1. Preparation of the Cu Complexes

Both the series,  $C_nCu$  and  $C_nOCu$  complexes were prepared according to Ohta *et al.*'s original publications [6, 8]. In  $C_nCu$  homologous series, the virgin crystals obtained by recrystallisation from organic solvent gave needle-like or cotton like crystals. However, the crystalline  $C_7Cu$  was obtained as flakes, which are totally different from the others.

# 2.2. X-ray Studies

Diffraction data on single crystals of  $C_6Cu$  (needle-like) and  $C_7Cu$  (flake-like) were collected at room temperature using a Rigaku AFC-5S four circle diffractometer with  $M_o-K_\alpha$  radiation. Single crystals of  $C_6Cu$  and  $C_7Cu$ , former recrystallised from ethyl acetate and the latter from hexane having the dimensions  $0.28 \, \text{mm} \times 0.16 \, \text{mm} \times 1.00 \, \text{mm}$  and  $0.36 \, \text{mm} \times 0.02 \, \text{mm} \times 1.00 \, \text{mm}$  respectively were used. Attempts at growing other crystals (other than  $C_8Cu/C_8OCu/C_{10}Cu$  studied by Usha *et al.*) of the  $C_nCu/C_nOCu$  series were not successful.

# 2.3. EPR Spectra

X and Q Band Spectra of the virgin samples (powders) were recorded with a Varian E 112 Spectrometer along with E257 Variable temperature accessory.

### 3. RESULTS AND DISCUSSION

# 3.1. X-ray Work on C<sub>6</sub>Cu and C<sub>7</sub>Cu

Tables II and III give the crystal data on the two complexes and Figures 2 and 3 give the molecular arrangements in the ab plane. Though C7Cu differs from C<sub>6</sub>Cu by a single CH<sub>2</sub> group and belongs to the same triclinic space group  $P\bar{1}$  with Z=1, the crystal structures and molecular arrangements are very different. Thus as seen from Tables II and III, 'c' axis in C<sub>6</sub>Cu has a value of 5.863 Å, whereas in  $C_7$ Cu, the value is almost double ~ 10.066 Å. Further, Figures 2 and 3 indicate that the molecular arrangements in 'ab' plane are very different, leading to an open structure in C<sub>6</sub>Cu (similar to Usha's C<sub>8</sub>Cu) but a rather crowded structure in C<sub>7</sub>Cu. In C<sub>6</sub>Cu, the molecules are more or less parallel to the ab plane with the alkyl chains in an all trans configuration. Further, there is no interaction between the organic groups viz., phenyl-chain, chain-chain, phenyl-phenyl etc., except the magnetic exchange interaction between Cu2+ ions on the inversion centre, along a column where Cu-Cu separation is  $\sim 5.863$  Å. But in C<sub>7</sub>Cu, the molecules are not parallel to the 'ab' plane but the phenyl rings with the chains tilt substantially from the 'ab' plane, causing the distance between two successive layers to increase, as observed. Of the four alkyl chains, only two are in an all trans form, while rest of the chains contain a cis part (circled in Fig. 3). Moreover, the stacking in the layer along 'ab' (2d) results in a very close interaction with the chains of a molecule in one corner with the chains of the molecule in the opposite corner of the unit cell. Further, four phenyl rings of two neighbouring molecules pile up along the b direction, possibly close enough (separation  $\sim 4.4 \,\text{Å}$ ) for  $\pi - \pi$  interaction as shown in Figure 3. Thus the molecules are not free in the 'ab' plane but are clubbed together forming interacting pairs. Thus C<sub>7</sub>Cu behaves differently, as columnar stacking in the 'c' direction is not possible, leading to an increase in the Cu-Cu distance (~10.066 Å), as compared to that in C<sub>8</sub>Cu. However, though the coordination around Cu is basically square planar in both C<sub>7</sub>Cu and C<sub>6</sub>Cu, the phenyl rings along with alkyl chains are severely distorted.

# TABLE II Crystal data for C<sub>6</sub>Cu

| 1.  | Empirical Formula                               | C(54) H(70) O(4) Cu(1)                  |
|-----|---|---|
| 2.  | Formula Weight                                  | 846.69                                  |
| 3.  | Crystal System                                  | Triclinic                               |
| 4.  | Lattice Parameters                              | a = 14.171 (6) angstroms                |
|     |   | b = 15.358 (4) angstroms                |
|     |   | c = 5.863 (3) angstroms                 |
|     |   | alpha = 95.20 (3) degrees               |
|     |   | beta = 93.15 (5) degrees                |
|     |   | gamma = 109.88 (3) degrees              |
|     |   | V = 1190.0 (8) angstroms**3             |
| 5.  | Space Group                                     | P-1(#2)                                 |
| 6.  | Z Value   | 1                                       |
| 7.  | Dcalc   | 1.18 g/cm**3                            |
| 8.  | FOOO  | 455                                     |
| 9.  | mu(Mo K-alpha)                                  | 5.00 cm** – 1                           |
| 10. | Diffractometer                                  | Rigaku AFC5S                            |
| 11. | Radiation                                       | Mo K-alpha (lambda = 0.71069) Graphite- |
|     |   | Monochromated                           |
| 12. | Temperature                                     | 23 degrees Cent.                        |
| 13. | 2-theta (max)                                   | 55.0 degrees                            |
| 14. | No. Observations $(I > 3.00 \text{ (sig }(I)))$ | 3130                                    |
| 15. | No. Variables                                   | 268                                     |
| 16. | Residuals: R; Rw                                | 0.046; 0.056                            |
| 17. | Goodness of Fit Indicator                       | 1.25                                    |
| 18. | Maximum Shift in Final Cycle                    | 1.36                                    |
| 19. | Largest Peak in Final Diff. Map                 | 0.56 e/angstrom**3                      |

# TABLE III Crystal data for C7Cu

| 1.  | Empirical Formula                       | C(58) H(78) O(4) Cu(1)                  |
|-----|---|---|
| 2.  | Formula Weight                          | 902.80                                  |
| 3.  | Crystal System                          | Triclinic                               |
| 4.  | Lattice Parameters                      | a = 10.316 (6) angstroms                |
|     |   | b = 13.326 (7) angstroms                |
|     |   | c = 10.066 (7) angstroms                |
|     |   | alpha = 90.77 (6) degrees               |
|     |   | beta = $92.71$ (7) degrees              |
|     |   | gamma = 71.25 (4) degrees               |
|     |   | V = 1309 (1)  angstroms**3              |
| 5.  | Space Group                             | P-1(#2)                                 |
| 6.  | Z Value                                 | 1                                       |
| 7.  | Deale                                   | 1.15 g/cm**3                            |
| 8.  | F000                                    | 487                                     |
| 9.  | Mu(Mo K-alpha)                          | 4.58 cm** - 1                           |
| 10. | Diffractometer                          | Rigaku AFC5S                            |
| 11. | Radiation                               | Mo K-alpha (lambda = 0.71069) Graphite- |
|     |   | monochromated                           |
| 12. | Temperature                             | 23 degrees Cent.                        |
| 13. | 2-theta(max)                            | 55.0 degrees                            |
| 14. | No. Observations $(I > 3.00 (sig (I)))$ | 1477                                    |
| 15  | No. variables                           | 286                                     |
| 16. | Residuals: R; Rw                        | 0.056; 0.064                            |
| 17. | Goodness of Fit Indicator               | 1.27                                    |
| 18. | Maximum Shift in Final Cycle            | 1.09                                    |
| 19. | Largest Peak in Final Diff. Map         | 0.53 e/angstrom**3                      |

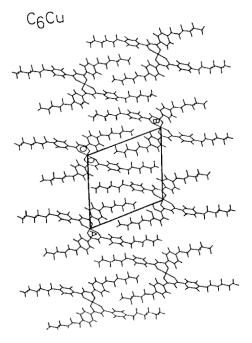


FIGURE 2 Molecular arrangement in the ab plane for C<sub>6</sub>Cu.

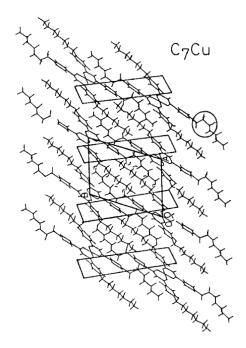


FIGURE 3 Molecular arrangement in the ab plane for C<sub>7</sub>Cu.

It has already been pointed out that Usha et al. [9] succeeded in obtaining two different crystallographic forms of  $C_8Cu$  viz., the more abundant needle-like form and the prismatic form (Cu-Cu distance  $\sim 10.34$  Å) having different crystal structures. It appears that  $C_7Cu$  has crystal parameters close to that of prismatic  $C_8Cu$  (Tab. IV), which again has been shown to be isomorphous with  $C_{10}$  Pd with a Pd-Pd distance  $\sim 10.26$  Å [20]. However, due to non-availability of sufficient quantity of prismatic crystals, it's properties could not be studied. In Table V are listed the average values of the bond lengths and valence angles characterizing the cores. No big difference can be seen from  $C_6Cu(K_1)$ ,  $C_7Cu(K_2)$ ,  $C_8Cu(N)$  and  $C_8Cu(P)$ . The core complex part  $CuO_4$  is completely planar. However, the torsion angles O-Cu-O-C are  $+3.1^\circ$  and  $-3.1^\circ$ . Hence the ligand part of the core bis (propane-1, 3-dionato) Cu(II) tilts up on one side and down on the other, to give an almost flat chair form.

### 4. EPR STUDIES AT X AND Q BANDS

EPR studies are recommended to be carried out at two different frequencies to arrive at the correct parameters. Classicaly EPR spectra was taken in the

|           |        | $C_6 - Cu(K_1)$ | $C_7 - Cu(K_2)$ | $C_8 - Cu(N)^*$ | $C_8 - Cu(P)^*$ |
|-----------|--------|-----------------|-----------------|-----------------|-----------------|
| a Ang     | stroms | 14.171(6)       | 10.316(6)       | 14.332(1)       | 11.621(2)       |
| b Ang     | stroms | 15.358(4)       | 13.326(7)       | 17.603(2)       | 12.817(2)       |
| _         | stroms | 05.863(3)       | 10.066(7)       | 05.821(3)       | 10.341(3)       |
|           | rees   | 95.20(3)        | 90.77(6)        | 98.86(2)        | 92.73(2)        |
|           | rees   | 90.15(7)        | 90.71(7)        | 93.58(3)        | 109.55(2)       |
|           | rees   | 109.88(3)       | 71.25(4)        | 107.35(1)       | 103.99(1)       |
| Space Gro |        | ₽Ĩ`´            | $P\bar{1}$      | $P\bar{1}$      | $P\bar{1}$      |
|           | 1      | 1               | 1               | 1               | 1               |

TABLE IV Crystal data for C<sub>n</sub>Cu

TABLE V Average dimensions of the core

|                      | $C_6-Cu(K_1)$ | $C_7 - Cu(K_2)$ | $C_8 - Cu(N)^*$ | $C_8 - Cu(P)^*$ |
|----------------------|---------------|-----------------|-----------------|-----------------|
| Cu-0/Å               | 1.905(2)      | 1.891(0)        | 1.908(6)        | 1.90(1)         |
| 0-C/A                | 1.276(3)      | 1.281(4)        | 1.27(1)         | 1.27(2)         |
| C-C/Å                | 1.394(4)      | 1.37(6)         | 1.389(2)        | 1.38(3)         |
| $0-Cu-0/^{\circ}$    | 92.87(9)      | 92.9(2)         | 92.8(2)         | 92.3(2)         |
| Cu-0-C/°             | 126.6(6)      | 126.7(8)        | 126.3(4)        | 127.2(2)        |
| 0-C-C/°              | 124.5(3)      | 124.3(3)        | 125.0(2)        | 124.2(4)        |
| $C - C - C/^{\circ}$ | 124.2(3)      | 124.2(3)        | 124.2(6)        | 124.4(6)        |

<sup>\*</sup> Ref. [23].

<sup>\*</sup> Ref. [23].

X band (9 GHz) for small molecules (inorganic and free radicals). But when signals were weak and proper resolution of the lines were not obtained, one went in for the Q band (35 GHz). At this higher frequency, not only were the lines more intense but also due to the spread out of the spectrum, many overlapping lines were clearly separated. In the present work also,  $C_n Cu$  $C_nOCu$  were first studied in the X band. In case of  $C_8Cu/C_{10}Cu$ ,  $g_{\parallel}$  was very weak but somewhat stronger in Q band. The exchange-narrowed  $g_{\perp}$  line was reasonably strong but was not much affected by change in frequency from X to Q band. However, in C<sub>8</sub>OCu (complex hyperfine structure), though more lines than the hyperfine quartet were evident in X band, they could not be resolved. But in Q band, the result was remarkable, with the overlapping lines separating out [15], which led to the inference that one was dealing with quadrupole forbidden transitions (q.f.t.), which was finally corroborated by computer simulation [16]. More recently, for large biological molecules, it has been seen [21] that somewhat better resolution of hyperfine structure is obtained at frequencies lower than X, viz., at L(1.2 GHz) and S(2-4 GHz) bands. However, in our case, the classical approach of going from X to Q band was highly satisfying. Spectra presented here are from powdered crystalline samples and their mesophases (Figs. 4-6) and the EPR parameters are given in Table VI.

# 4.1. Studies in C<sub>n</sub>Cu System

Work was extended to cover C<sub>6</sub>Cu, C<sub>7</sub>Cu, C<sub>9</sub>Cu and C<sub>12</sub>Cu (C<sub>8</sub>Cu and C<sub>10</sub>Cu was reported before). C<sub>6</sub>Cu behaved as the prototype columnar one dimensional linear chain, as in C<sub>8</sub>Cu and gave the single highly asymmetric exchange-narrowed  $g_{\perp}$  line, along with a weak  $g_{\parallel}$ .  $C_{12}Cu$  also showed a strongly exchange-narrowed asymmetric  $g_{\perp}$  line ( $\Delta H \sim 20$  Oe) but the weak  $g_{\parallel}$  was so broad, that it was followed immediately by  $g_{\perp}$ . Interestingly, in the C<sub>n</sub>Cu series, there seems to be a sort of odd-even effect (Fig. 4a). Thus C<sub>7</sub>Cu and C<sub>9</sub>Cu show a different spectra viz.,  $g_{\parallel}$  is a well separated hyperfine quartet  $(A_{\parallel} \sim 180 \,\text{Oe})$  for both, and  $g_{\perp}$  is still exchange-narrowed in C<sub>9</sub>Cu but not in  $C_7$ Cu. The  $g_{\perp}$  line-width of the interacting pair in Q band at room temperature is  $\sim 130 \, \text{Oe}$  in the former. Unfortunately for C<sub>9</sub>Cu, which is rather unique in having three different mesophases, there is no crystal structure data. It also shows the copper hyperfine quartet in  $g_1$  but  $g_{\perp}$  is now splitted up giving  $g_2$  and  $g_3$ ,  $g_2$ , which corresponds to  $g_{\perp}$  is also exchangenarrowed ( $\Delta H \sim 25$  Oe) and  $g_3$  is weak, showing that the system is non-axial and thus different from other members of the C<sub>n</sub>Cu series. The texture in the

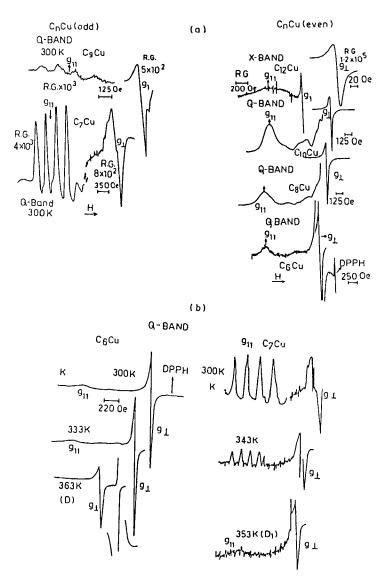


FIGURE 4 (a) Room temperature EPR spectra of  $C_nCu$ : Odd  $C_7Cu$  and  $C_9Cu$ , even  $C_6Cu$ ,  $C_8Cu$ ,  $C_{10}Cu$  and  $C_{12}Cu$ ; (b) Temperature variation studies of  $C_6Cu$  and  $C_7Cu$  in Q Band.

 $D_1$  phase is spiral, which may possibly result from a helicoidal packing in the K phase, as has been observed in Hexa-hexyl-thiotriphenylene [22] and mosaic in  $D_2$ , which wrinkles in  $D_3$ .

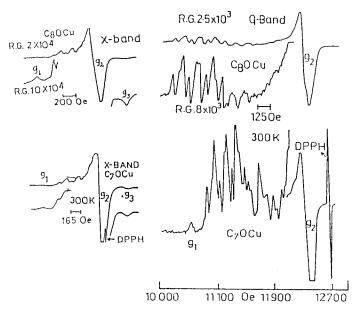


FIGURE 5 X/Q Band room temperature spectra of C<sub>n</sub>OCu; C<sub>7</sub>OCu and C<sub>8</sub>OCu.

# 4.2. Temperature Variation Studies in C<sub>6</sub>Cu and C<sub>7</sub>Cu

EPR spectra in Q band (Fig. 4b) are presented for C<sub>6</sub>Cu/C<sub>7</sub>Cu from K to the discotic phases. The exchange-narrowed g<sub>1</sub> line in C<sub>6</sub>Cu becomes wider (decreases in intensity) with increasing temperature and  $g_{\parallel}$  intensity also diminishes, till at the mesophase,  $g_{\parallel}$  almost vanishes. In C<sub>7</sub>Cu the effects are striking. The quartet in the solid is slightly distorted but with temperature increasing, the distortion disappears and in the  $D_1$  phase (80°C) the quartet vanishes to give a weak  $g_{\parallel}$ . Thus  $g_{\perp}$  line narrows from 130 Oe in K phase to 50 Oe in  $D_1$ . There is no significant change in the spectra from  $D_1$  to  $D_2$ phase (100°C) except a marginal narrowing ( $\Delta H \sim 45$  Oe) of  $g_{\perp}$ . However, in C<sub>7</sub>Cu, the intermolecular interactions in the K phase is not so strong as in dimer C<sub>8</sub>OCu. The weak nature of the interactions among adjacent neighbours is clearly shown up on heating. Thus at the mesophase, these interactions disappear (Fig. 4b), leading to the collapse of the hyperfine quartet, which is not the case in dimeric C<sub>8</sub>OCu. Possibly, close packing in this phase leads to a larger exchange interaction, which is reflected in the narrowing of the  $g_{\perp}$  from 130 Oe to 50 Oe in C<sub>7</sub>Cu. This is opposite to that in C<sub>6</sub>Cu, where the exchange-narrowed line in K phase (30 Oe) increases to 60 Oe at 140°C (mesophase).

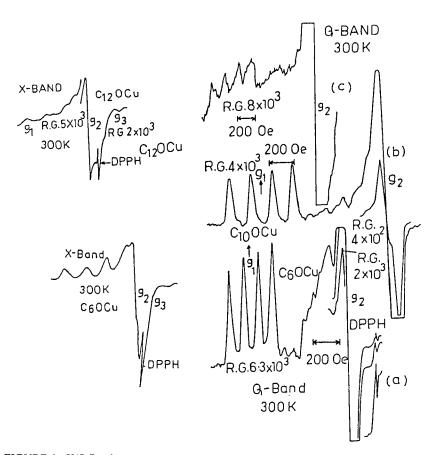


FIGURE 6 X/Q Band room temperature spectra of C<sub>n</sub>OCu; C<sub>6</sub>OCu, C<sub>10</sub>OCu and C<sub>12</sub>OCu.

The difference in the behaviour of  $C_6Cu$  (even) and  $C_7Cu$  (odd) is also reflected in the dynamics of the two systems. Thus <sup>1</sup>H (proton) relaxation time  $T_1$  at room temperature is very short in the exchange-narrowed  $C_6Cu(\sim 1.9 \, \mathrm{msecs})$ , comparable to  $T_1$  in  $C_8Cu$  ( $\sim 2 \, \mathrm{msec}$ ), but very different from the rather long  $T_1 \sim 68.9 \, \mathrm{msecs}$  in  $C_7Cu$  (unpublished work). However,  $T_1$  has not been measured in  $C_9Cu$ .

# 4.3. Studies in CnOCu System

Detailed EPR studies on C<sub>8</sub>OCu has already been reported by us [18]. Computer-simulation has confirmed that the complex hyperfine structure

TABLE VI EPR parameters for C<sub>n</sub>Cu and C<sub>n</sub>OCu complexes

| $C_nCu$  | Band       | 8.<br>  8 | 82<br>8± | 83    | $\Delta H Oe$    | $A_{\parallel}$ | C,OCu                   | Band      | 81    | 82    | 83    | $\Delta H_{\perp}$ | 90                |
|--|------------|-----------|----------|-------|------------------|-----------------|-------------------------|-----------|-------|-------|-------|--------------------|-------------------|
| C <sub>6</sub> Cu(RT)  | 0          | 2.283     | 2.068    |       | 9<br>9<br>9<br>9 |                 | C,OCu(RT)               | o×        | 2.263 | 2.056 | 1.855 | 120                | 08<br>8<br>8<br>8 |
| C,Cu(RT)   | 0          | 2.241     | 2.051    |       | 130              | 180             | C <sub>7</sub> OCu(RT)  | 0         | 2.366 | 2.059 |       | 130                | 180               |
| $D_1$  | ,          | 2.245     | 2.055    |       | 20               |                 |                         | <b>′×</b> | 2.625 | 2.040 | 1.779 | 125                |                   |
| $D_2$  |            | 2.245     | 2.055    |       | 45               |                 |                         |           |       |       |       |                    |                   |
| C <sub>s</sub> Cu(RT)  | 0          | 2.212     | 2.053    |       | 30               |                 | C <sub>8</sub> OCu(RT)  | ×         | 2.55  | 2.060 | 1.78  | 140                |                   |
| $D_1$  |            | 2.214     | 2.051    |       | 37.5             |                 | M                       |           | 2.46  | 2.060 | 1.90  | 140                |                   |
| $D_2$  |            | 2.215     | 2.050    |       | 40               |                 |                         |           |       |       |       |                    |                   |
| RT   | ×          | 2.243     | 5.069    |       | 70               |                 | RT                      | 0         | 2.377 | 2.252 | 1     | 140                |                   |
| $D_1$  |            | 2.240     | 2.070    |       | 37.5             |                 | M                       |           | 2.102 | 2.092 |       | 125                |                   |
| $D_2$  |            | 2.240     | 2.070    |       | 40               |                 |                         |           |       |       |       |                    |                   |
| C <sub>2</sub> Cu(RT)  | 0          | 2.254     | 2.057    | 2.044 | 25               | 180             |                         |           |       |       |       |                    |                   |
| CloCu(RT)  | <b>'</b> × | 2.280     | 2.061    |       | 70               |                 | C <sub>10</sub> OCu(RT) | 0         | 2.25  | 2.056 |       | 140                | 180               |
| Q  |            | 2.278     | 2.060    |       |                  |                 |                         |           |       |       |       |                    |                   |
| RT   | 0          | 2.380     | 2.061    |       | 30               |                 |                         |           |       |       |       |                    |                   |
| D  | 0          | 2.382     | 2.059    |       |                  |                 |                         |           |       |       |       |                    |                   |
| $C_{12}C_{\omega}(RT)$   | ×          | 2.49      | 1.886    |       | 20               |                 | C <sub>12</sub> OCu(RT) | 0         |       | 2.054 |       | 80                 | 180               |
|  |            |           |          |       |                  |                 | M                       |           |       | 5.05  |       | 2                  |                   |
|  |            |           |          |       |                  |                 | /<br>Td                 | >         | 2.410 | 2.054 | 1 845 | 5 5                | 133               |
| C <sub>8</sub> Cu in C <sub>8</sub> Pd   | 0          | 2.258     | 2.138    | 2.03  |                  | 180             | ž                       | •         | 71.41 | 10.7  | 60.1  | <b>;</b>           |                   |
| C10Cu in C10Pd   | 0          | 2.234     | 2.043    | 2.038 |                  | 180             |                         |           |       |       |       |                    |                   |
| $\Delta H_{\perp}$ = line width.<br>$A_{\parallel}$ = hyperfine interaction.<br>M = Mesophase.<br>I = Isotropic Phase.<br>I = Ref. [19]. |            |           |          |       |                  |                 |                         |           |       |       |       |                    |                   |

arises from overlapping Cu quartets and quadrupole forbidden transitions from  $\Delta M_1 = \pm 1$  and  $\pm 2$ . Usha's X-ray studies [12] have shown that  $C_8OCu$  is a dimer, with no centre of invension. In the present work, of the other  $C_nOCu$  complexes studied, only  $C_7OCu$  shows a complex hyperfine spectrum, very similar to that of  $C_8OCu$ . However, there is no X-ray data (no single crystal could be grown) to conclude that it is a dimer like  $C_8OCu$ . However, from the similarity in the EPR spectra of the two, it may not be unreasonable to infer that it is also in all possibility a dimer.

It may be pointed out that EPR spectra throughout the series is in harmony with the non-axial nature of  $C_nOCu$ . But  $C_6OCu$ ,  $C_{10}OCu$  and  $C_{12}OCu$  exhibit only the Cu hyperfine (hf.) quartets but no quadrupolar forbidden transitions (q.f.t). For the first two, this quartet structure is in  $g_1$ , but in  $C_{12}OCu$ , which is somewhat unique, the hf. quartet is in between  $g_1$  and  $g_2$ . However,  $g_2$  lines for all three reveal weak or no exchange; asymmetries and line widths indicate, that exchange if present, is nowhere as strong as in the  $C_nCu$  series (Tab. VI). The absence of q.f.t. in these three, indicate that these are monomers with a more symmetrical structure, as compared to  $C_8OCu/C_7OCu$ . However, again there is no crystal structure data to support our contention.

Anisotropy is a primary criterion in liquid crystal formation and survey of the literature shows that the alkyl chain  $C_8H_{17}$  is possibly the most favoured chain in discotic and other liquid crystals. Possibly, this chain length is ideal for packing considerations, particularly, for the trans configuration of the chain in Ohta's model [6], based on  $C_8Cu$ . However, when n > 8, the trans configuration can lead to very loose packing of the molecules in the solid (thermodynamically not the lowest energy configuration), so that the conformation from n > 8 may be different. Thus the chains may be alternately up and down the core with a cis conformation, not only in  $C_{12}OCu$ ,  $C_{10}OCu$  and  $C_6OCu$ , but possibly in  $C_9OCu$  and  $C_{11}OCu$ . This structure has a high symmetry, compared to  $C_7OCu$  and  $C_8OCu$ , in which the trans alkyl chains in the plane of the core, leads to a highly asymmetric structure, responsible for the quadrupolar field gradient (q). For n > 8 (symmetric lamellar), with the higher symmetry, 'q' would tend to zero, so no q.f.t. are observed.

Further, temperature variation studies of  $C_{12}OCu$  reveal that hyperfine quartets virtually disappear in the mesophase but reappears in the isotropic phase, albeit with a lower intensity and a somewhat smaller hyperfine interaction constant  $A_{\parallel}$ . However, it has been pointed out that the quartets do not vanish in the mesophase of  $C_8OCu$ . Thus the collapse of the hyperfine quartet in the mesophase of  $C_{12}OCu$ , further supports our

contention that the chain conformation along with the packing is different from that of  $C_8OCu$ . This is also reflected in the shape of  $C_{12}OCu$ , which is needle like, being different from other rod or plate shaped members of  $C_nOCu$  series.

# 4.4. Effect of Frequency and Temperature on Line Widths ( $\Delta H$ )

As  $g_{\parallel}$  lines in many cases are rather weak or split (quartet), one focusses on the  $g_{\perp}$  line-width. In the  $C_nCu$  series, the exchange-narrowed line-widths ( $\Delta H$ ) in  $C_8Cu/C_{10}Cu$  are the same in the K phase viz.,  $\sim 20$  Oe in X band and  $\sim 30$  Oe in Q band (Tab. VI). However,  $\Delta H$  changes sharply in the mesophase due to disordering. For  $C_8Cu$ ,  $\Delta H$  values ( $\sim 37.5$  Oe in  $D_1$  and  $\sim 40$  Oe for  $D_2$ ) are independent of frequency within experimental error. This is not the case in  $C_{10}Cu$ , as D has a value of  $\sim 35$  Oe in X band but 60 Oe in Q band. Possibly, the exchange interaction in  $C_{10}Cu$  is somewhat weaker than in  $C_8Cu$ . Thus as expected, in case of the exchange-narrowed lines in the crystalline phase, change of frequency strongly affects  $\Delta H$ . Increasing the temperature, leads to the mesophases, wherein,  $\Delta H$  is found to be much larger, as disordering affects exchange.

In  $C_8OCu$ , where  $\Delta H$  is not exchange-narrowed, frequency seems to have no effect in the K phase, as both in X and Q bands,  $\Delta H \sim 140$  Oe. Spin – spin interaction in  $C_8OCu$  dimer possibly determine  $\Delta H$ . However, in the mesophase, the higher magnetic field in Q band (12,500 Oe) possibly breaks up the dimer, [15] as evidenced in the collapse of q.f.t. This shows a  $\Delta H \sim 125$  Oe, whereas in the X band (3000 Oe), the hf. structure has just started to distort, but  $\Delta H$  remains unaltered ( $\sim 140$  Oe).

Not unexpectedly,  $A_{\parallel}$  in the hyperfine quartet in both  $C_nCu$  and  $C_nOCu$  is almost constant (~180 Oe) within experimental error. This hf. interaction constant reflects the bonding situation in the  $CuO_4$  core, which is essentially the same in these square planar complexes and is unaffected by the alkyl chain length, as also packing. Thus, even in dilute systems viz.,  $C_8Cu/C_{10}Cu$  doped in  $C_8Pd/C_{10}Pd$  respectively,  $A_{\parallel}$  remains unchanged, as it is a molecular parameter.

### 5. CONCLUSIONS

EPR studies in  $C_nCu/C_nOCu$  series reveal basically three different spectral types in powders:

- 1. Single strongly exchange-narrowed asymmetric  $g_{\perp}$  line, along with a weak  $g_{\parallel}$  for C<sub>6</sub>Cu, C<sub>8</sub>Cu, C<sub>10</sub>Cu and C<sub>12</sub>Cu (even) but not for C<sub>7</sub>Cu and C<sub>9</sub>Cu (odd) in the columnar C<sub>n</sub>Cu series.
- 2. Normal hyperfine quartet for copper in  $g_1$  with a single  $g_2$  and a weak  $g_3$  for C<sub>6</sub>OCu and  $C_{10}$ OCu but in between  $g_1$  and  $g_2$  in C<sub>12</sub>OCu in the lamellar C<sub>n</sub>OCu series.
- 3. Hyperfine quartet with quadrupole forbidden transitions (q.f.t.) in between weak  $g_1$  and the strong broad single  $g_2$ , along with a weak  $g_3$  in  $C_8OCu$  and  $C_7OCu$ . In  $C_8OCu$ , dimer formation (confirmed from X-ray) is possibly responsible for the forbidden transitions, which however vanish at the mesophase, but the quartet persists.
- 4.  $C_7Cu$  and  $C_9Cu$  in the  $C_nCu$  series are rather unique, as they do not exhibit a columnar structure. Both give a hyperfine quartet in  $g_{\parallel}$ .  $C_7Cu$  like  $C_8Cu$ , give two mesophases, but unlike the latter, do not give an exchanged-narrowed  $g_{\perp}$ , as molecular arrangement within the crystal is very different. Thus though z=1 for both, interaction between two adjacent molecules in  $C_7Cu$  along end-chains, as also phenyl-phenyl interaction, cause them to behave as a pair in the crystalline state, instead of the linear Heisenberg chain in columnar  $C_8Cu$ . However, this interaction is weak and vanishes in the mesophase (unlike the dimeric  $C_8OCu$ ), as manifested in the disappearance of the quartet to give a weak  $g_{\parallel}$ .  $C_9Cu$  is the only member which shows three mesophases and may have a helicoidal structure, at least, not the regular columnar structure of the  $C_nCu$  series. EPR spectra shows  $C_9Cu$  to be the only non-axial member in this series.

Thus X-ray and EPR studies indicate that within each series, some members exhibit anomalous behaviour viz.,  $C_7Cu$  and  $C_9Cu$  in  $C_nCu$  and  $C_7OCu/C_8OCu$  in  $C_nOCu$ . Further, weak or no exchange in  $C_7Cu$  is confirmed from the rather long proton ( $^1H$ ) relaxation time (NMR  $T_1 \sim 68.9$  msecs.), as compared to the short relaxation time  $T_1 \sim 1.9$  msecs. in the strongly exchange-narrowed  $C_6Cu$ . The above studies along with melting points and miscibility behaviour, indicate that chain length does play an important role in both the series but not in a linear fashion. Interestingly, all members of the  $C_nCu$  and  $C_nOCu$  studied so far [23] crystallise in the triclinic space group ( $P\bar{1}$ ), which possibly satisfies the structural requirement for efficient packing.

# Acknowledgements

Our thanks are due to Prof. S. Subramanian, Head, RSIC of IIT Madras for his kind cooperation, and to Dr. Srilekha Banerjee for her help during the analysis of the EPR spectra. Monisha Bose and C. K. Majumdar are grateful to the DST, Government of India for sanctioning the project on 'Metallomesogens'.

### References

- [1] A. M. Giroud-Godquin and J. Billard, Mol. Cryst. Liq. Cryst., 66, 147 (1981).
- [2] A. M. Levelut, J. Chim. Physique, 80, 149 (1983).
- [3] J. Billard, C. R. Acad Sc. Paris, Series II, 299, 905 (1984).
- [4] K. Ohta, A. Ishii, I. Yamamoto and K. Matsuzaki, J. Chem. Soc. Commun., p. 1099 (1984).
- [5] K. Ohta, A. Ishii, H. Muroki, I. Yamamoto and K. Matsuzaki, Mol. Crys. Liq. Cryst., 116, 299 (1985).
- [6] K. Ohta, H. Muroki, A. Takagi, K. Hatada, H. Ema, I. Yamamoto and K. Matsuzaki, Mol. Cyst. Liq. Cryst., 140, 131 (1986).
- [7] H. Sakashita, A. Nishitani, Y. Sumiya, H. Terauchi, K. Ohta and I. Yamamoto, Mol. Cryst. Lig. Cryst., 163, 211 (1988).
- [8] K. Ohta, Hiromitsu Muroki, Akira Takagi, Iwao Yamamoto and Kei Matszaki, Mol. Cryst. Liq. Cryst., 135, 247 (1986).
- [9] K. Usha and K. Vijayan, Mol. Cryst. Liq. Cryst., 220, 77 (1992).
- [10] A. M. Giroud-Godquin and J. Billard, Mol. Cryst. Liq. Cryst., 97, 287 (1983).
- [11] R. Fugnitto, H. Strzelecka, A. Zann, J. C. Dubois and J. Billard, J. Chem. Soc. Chem. Comm., p. 271 (1980).
- [12] K. Usha and K. Vijayan, Mol. Cryst. Liq. Cryst., 174, 39 (1989).
- [13] M. P. Eastman, M. L. Horng, B. Freiha and K. W. Sheu, Liq. Cryst., 2, 223 (1987).
- [14] Monisha Bose and B. K. Sadashiva, Mol. Cryst. Liq. Cryst. Letters, 8, 59 (1991).
- [15] Monisha Bose and B. K.Sadashiva, Mol. Cryst. Liq. Cryst. Letters, 8, 137 (1992).
- [16] Monisha Bose, Jayashree Saha, Chanchal K. Majumdar and B. K. Sadashiva, Mol. Cryst. Lig. Cryst., 307, 43 (1997).
- [17] K. Usha, K. Vijayan and B. K. Sadashiva, Mol. Cryst. Liq. Cryst., 201, 13 (1991).
- [18] B. Bleaney, *Phil. Mag.*, **42**, 441 (1951).
- [19] Monisha Bose, Chanchal K. Majumdar and B. K. Sadashiva, Mol. Cryst. Liq. Cryst., 307, 57 (1997).
- [20] K. Usha, K. Vijayan, B. K. Sadashiva and P. R. Rao, Mol. Cryst. Liq. Cryst., 185, 1 (1990).
- [21] J. S. Hyde and W. Froncisz, Ann. Rev. Biophys. Bioeng., 11, 391 (1982).
- [22] E. Fontes, P. A. Heine and W. H. de Jeu, Phys. Rev. Letters, 61, 1202 (1988).
- [23] K. Usha, K. Vijayan and S. Chandrasekhar, Liq. Cryst., 15, 575 (1993).